

Visa Credit Card Number Generator Document

visa credit card number generator has become a topic of interest for various reasons, ranging from testing payment systems to understanding the structure of credit card numbers. In this comprehensive guide, we will explore what a visa credit card number generator is, how it works, its legitimate uses, potential risks, and how to ensure safety and compliance when dealing with such tools. Whether you're a developer, a cybersecurity professional, or simply curious about credit card number generation, this article provides valuable insights into the subject.

Understanding the Visa Credit Card Number Generator

What Is a Visa Credit Card Number Generator?

A visa credit card number generator is a software tool or online service that produces random or pseudo-random credit card numbers that mimic genuine Visa cards. These generated numbers follow the structural patterns and checksum algorithms of real Visa card numbers but are not linked to actual bank accounts or financial institutions.

How Does It Work?

Visa card numbers are generated based on the ISO/IEC 7812 standard, which defines the structure of identification numbers for payment cards. The generator typically:

- Randomly or systematically creates a card number sequence
- Ensures the number adheres to the Visa format (starting with a '4')
- Calculates and appends a valid Luhn checksum digit to make the number appear legitimate
- May include other data like expiration dates, CVV codes, and cardholder names for testing purposes

Structure of a Visa Card Number

Understanding the structure helps in grasping how generators produce plausible numbers:

- **Major Industry Identifier (MII):** The first digit '4' indicates it's a Visa card.
- **Issuer Identification Number (IIN):** The next 5 digits specify the issuing bank or institution.
- **Account Number:** The following 9 digits uniquely identify the cardholder's account.
- **Checksum Digit:** The final digit is calculated using the Luhn algorithm to validate the entire

number.

Most generators produce numbers that replicate this structure, making them appear authentic for testing or educational purposes.

Legitimate Uses of a Visa Credit Card Number Generator

While generating fake credit card numbers may seem suspicious, there are several legitimate and ethical applications:

1. Testing Payment Processing Systems

Developers and QA teams use generated card numbers to test online payment gateways, checkout processes, and fraud detection systems without risking real funds or sensitive data.

2. Software Development and Debugging

When creating financial applications, developers need a reliable set of test data that simulates real-world scenarios, which generated card numbers provide.

3. Educational Purposes

Students and cybersecurity professionals study the structure, validation algorithms, and security aspects of credit cards using generated numbers to understand how payment systems work.

4. Compliance with PCI DSS Standards

Organizations in the payment industry utilize test card numbers compliant with PCI DSS standards for internal testing, ensuring they do not store or transmit real card data.

Risks and Ethical Considerations

Using or creating fake credit card numbers for malicious purposes is illegal and unethical. Here are some points to consider:

- **Fraudulent Activities:** Using generated or stolen card numbers for unauthorized transactions is criminal behavior.
- **Legal Implications:** Generating or using fake credit card numbers outside legitimate testing contexts can lead to legal penalties.
- **Security Risks:** Relying on untrusted generators may expose users to malware or scams.

- **Data Privacy:** Generating and storing fake card data should comply with data privacy laws and standards.

Always ensure that the use of such tools is confined to ethical, legal, and approved environments.

How to Use a Visa Credit Card Number Generator Responsibly

If you decide to use a visa credit card number generator for legitimate purposes, follow these best practices:

1. Verify the Source

Choose reputable and well-reviewed generators that produce compliant test data. Avoid suspicious or untrusted websites.

2. Understand the Limitations

Generated numbers are not real and cannot be used for actual transactions. They are solely for testing or educational purposes.

3. Respect Privacy and Security Standards

Ensure your testing environments are secure, and avoid using generated data in live production systems.

4. Use for Testing Only

Never attempt to process transactions with generated numbers outside authorized testing environments.

5. Keep Up with Industry Standards

Stay updated on PCI DSS guidelines and other compliance protocols to ensure your testing practices remain legal and secure.

Popular Tools and Resources for Generating Visa Card Numbers

There are numerous tools available online, some free and others paid, that facilitate the generation of Visa credit card numbers:

- **Online Card Number Generators:** Websites offering quick, easy-to-use generators for testing purposes.
- **API-Based Generators:** Services that provide APIs for integrating card number generation into software applications.
- **Open-Source Libraries:** Programming libraries in languages like Python, JavaScript, or Java that allow developers to generate card numbers programmatically.

When selecting a tool, ensure it complies with industry standards and respects user privacy.

Understanding the Limitations of Card Number Generators

While helpful for testing, card number generators have limitations:

- **Not linked to real accounts:** Generated numbers are invalid for actual transactions.
- **Cannot be used for fraud:** They are designed for testing, not for illegal activities.
- **Potential for misuse:** Unauthorized use can lead to legal consequences.
- **Limited data:** Many generators do not provide additional card details like CVV, expiration date, or billing address unless explicitly generated.

Always recognize that the primary purpose of such tools is testing and education.

Conclusion

A **visa credit card number generator** serves as a valuable tool in various legitimate contexts, especially for developers, testers, and cybersecurity professionals. Understanding the structure and algorithms behind Visa card numbers, such as the Luhn checksum, helps in appreciating how these generators produce plausible data. However, it is crucial to use these tools responsibly, ethically, and within the bounds of the law. Misuse or malicious intent can lead to serious legal repercussions and security issues. Always prioritize security standards, verify the legitimacy of tools, and ensure your activities align with industry regulations.

By leveraging the power of a Visa credit card number generator ethically, you can streamline testing processes, enhance your understanding of payment systems, and contribute to safer digital financial environments.

Visa credit card number generator has become a topic of interest for various reasons, ranging from software testing and development to understanding the structure of credit card numbers for educational purposes. While generating fake credit card numbers can be useful in controlled environments, it's crucial to emphasize that using or distributing such numbers for fraudulent activities is illegal and unethical. This guide aims to provide a comprehensive overview of what a

Visa credit card number generator is, how Visa card numbers are structured, the importance of validation algorithms, and best practices for ethical and legal use.

Understanding the Basics of Visa Credit Card Numbers

What Is a Visa Credit Card Number?

A Visa credit card number generator is a tool or algorithm designed to produce random or pseudo-random sequences of digits that resemble real Visa credit card numbers. These generated numbers typically mimic the structure and validation rules of genuine Visa cards but are not linked to actual accounts or financial institutions.

Why Would Someone Use a Visa Credit Card Number Generator?

- Software Testing: Developers working on payment processing systems need to test their applications without risking real financial data.
- Educational Purposes: Teaching students about the structure of credit card numbers and validation algorithms.
- Security Research: Understanding vulnerabilities in card number validation or fraud detection systems.

The Structure of Visa Credit Card Numbers

Visa credit card numbers follow a specific format defined by the ISO/IEC 7812 standard. Understanding this structure is essential for generating realistic numbers.

Key Components of a Visa Card Number

- Issuer Identification Number (IIN): The first 6 digits of the card number identify the issuing bank or institution. For Visa, the IIN typically starts with a '4'.
- Account Number: The subsequent digits (usually 6-12 digits) identify the individual account.
- Check Digit: The final digit of the card number is the check digit, calculated using the Luhn algorithm to validate the number.

Visa Card Number Format Summary

- Length: 13 to 19 digits (most commonly 16)
- Starts with: '4'

- IIN: 4-6 digits (e.g., '4' or '400000')
- Remaining digits: Customer account number
- Check digit: Last digit, computed via Luhn algorithm

The Role of the Luhn Algorithm in Validation

What Is the Luhn Algorithm?

The Luhn algorithm (also known as the "modulus 10" algorithm) is a checksum formula used to validate various identification numbers, including credit card numbers. It helps detect accidental errors like mistyped digits.

How Does the Luhn Algorithm Work?

- Starting from the rightmost digit (excluding the check digit), double every second digit.
- If doubling results in a number greater than 9, subtract 9 from it.
- Sum all the digits (both modified and unmodified).
- The total sum should be divisible by 10 for the number to be valid.

Importance in Card Number Generation

When creating a fake card number, the generator must produce a number that passes the Luhn check to appear authentic and be accepted by validation systems during testing.

How to Generate a Valid Visa Credit Card Number

Step-by-Step Process

- Select the IIN (Issuer Identification Number):
 - For Visa, start with '4.'
 - Example: '4' or '400000'
- Generate the Account Number:
 - Randomly generate the necessary digits to reach the desired total length (e.g., total length of 16

digits).

- Ensure the total length is appropriate (commonly 16 digits for Visa).
- Calculate the Check Digit:
 - Use the Luhn algorithm to determine the last digit.
 - This involves computing the check digit based on the preceding digits.
- Combine All Components:
 - Concatenate the IIN, account number, and check digit to form the complete card number.

Example: Generating a 16-Digit Visa Card Number

Suppose we choose:

- IIN: 4 (single digit for simplicity)
- Account number: 15 digits (to reach 16 total digits)

Generate random digits for the account number:

- Account number: 1234 5678 9012 345

Calculate the check digit for the sequence '4 1234 5678 9012 345' (excluding the last digit). Once obtained, append the check digit.

Ethical and Legal Considerations

While tools like visa credit card number generators can be useful for legitimate testing and educational purposes, it is vital to adhere to ethical guidelines:

- Do Not Use for Fraud: Never attempt to use generated card numbers for unauthorized transactions.
- Limit Usage to Testing Environments: Use in isolated, controlled environments where no real

financial data is involved.

- Respect Privacy and Laws: Unauthorized use of real or fake credit card data can lead to criminal charges.

Tools and Resources for Generating Visa Card Numbers

Popular Card Number Generators

Many online tools and software libraries can generate valid-looking Visa card numbers:

- Open-source libraries: Available in languages like Python, JavaScript, etc.
- Online generators: Websites that produce test card numbers for various card brands.

Example Libraries

- Python: ``faker`` library with credit card provider options.
- JavaScript: ``creditcard-generator`` npm package.

Implementation Tips

- Always ensure the generated number passes the Luhn check.
- Use a fixed IIN when testing specific scenarios.
- Avoid exposing or storing generated numbers insecurely.

Best Practices for Using Generated Visa Credit Card Numbers

- Use for Testing Only: Never attempt to process transactions with generated numbers.
- Validate with Luhn Algorithm: Always verify that the generated number passes the checksum.
- Document Usage: Keep records of generated test data for troubleshooting.
- Stay Updated: Card number standards evolve; ensure your generator aligns with current standards.

Conclusion

The visa credit card number generator is a valuable tool for developers, educators, and security researchers when used responsibly. Understanding the structure of Visa card numbers, the role of the Luhn algorithm, and ethical considerations is essential for leveraging these tools correctly.

Remember, authenticating the validity of generated numbers through checksum algorithms ensures realistic testing scenarios, but always prioritize legal and ethical boundaries in your work.

Disclaimer: This guide is intended for educational and authorized testing purposes only. Do not use generated credit card numbers for illegal activities.

Question Is it legal to use a visa credit card number generator online? No, using a visa credit card number generator for fraudulent purposes is illegal and can lead to serious legal consequences. These tools are often used for testing or educational purposes only. **Can a visa credit card number generator be used for online purchases?** Typically, no. Legitimate merchants require real, authorized credit card information for transactions. Generators do not produce valid, authorized card numbers for actual purchases. **How do credit card number generators work?** Credit card number generators create numbers based on algorithms that mimic the structure of real credit cards, including valid prefixes and checksum algorithms, but they do not generate authorized or real card data. **Are there risks associated with using a visa credit card number generator?** Yes, using such generators can expose you to malware, scams, or legal issues if used improperly. They should only be used for testing purposes in secure, authorized environments. **What are legitimate alternatives to test online payment systems without using real credit card data?** Legitimate options include using sandbox environments provided by payment gateway services, or utilizing test credit card numbers provided by payment processors for development and testing purposes.

Related keywords: credit card generator, fake credit card numbers, visa card generator, credit card spoofing, test credit card numbers, dummy credit card data, generate visa card info, credit card number creator, visa card fake generator, credit card number tool

chemistry worksheet atomic number and mass number

Chemistry worksheet atomic number and mass number are fundamental concepts in the study of chemistry that help students and enthusiasts understand the structure of atoms. These concepts are essential for grasping how elements are organized, how isotopes differ from each other, and how atomic properties influence chemical behavior. A well-designed chemistry worksheet focusing on atomic number and mass number serves as an excellent educational tool to reinforce these foundational ideas, making complex topics more accessible through practical exercises, clear explanations, and engaging activities.

Understanding Atomic Number and Mass Number

What is the Atomic Number?

The atomic number (denoted as Z) is a fundamental property of an element. It represents the number of protons present in the nucleus of an atom of that element. Since protons are positively charged particles, the atomic number essentially defines the element itself because each element has a unique number of protons.

- Determines the element's identity (e.g., Hydrogen has $Z=1$, Carbon $Z=6$).
- Impacts the element's chemical properties and reactivity.
- Defines the number of electrons in a neutral atom.

What is the Mass Number?

The mass number (denoted as A) refers to the total number of protons and neutrons in an atom's nucleus. Unlike the atomic number, which is unique for each element, the mass number can vary among isotopes of the same element.

- Calculates the approximate atomic mass of an isotope.
- Used to distinguish between different isotopes.
- Not a whole number for elements with naturally occurring isotopic mixtures, where average atomic mass is used instead.

Key Differences Between Atomic Number and Mass Number

Feature	Atomic Number (Z)	Mass Number (A)
Definition	Number of protons in an atom	Total number of protons and neutrons
Symbol	Z	A
Unique for each element	Yes	No (varies among isotopes)
Determines	Element identity	Isotope type

Using a Chemistry Worksheet to Learn Atomic Structure

A chemistry worksheet centered on atomic number and mass number provides structured exercises that enhance understanding through various activities such as:

- Identifying elements based on atomic number
- Calculating the number of neutrons in isotopes
- Determining the isotope symbol notation
- Matching isotopes to their atomic and mass numbers
- Understanding the concept of atomic mass as an average of isotopes

These activities help students master key concepts and develop problem-solving skills essential for advanced chemistry topics.

How to Calculate Atomic Number and Mass Number

Calculating Atomic Number

The atomic number is usually given or can be found on the periodic table. When not provided directly, it is often determined by:

- Looking at the element symbol (e.g., C for Carbon, atomic number 6)
- Using the periodic table
- Counting the number of protons in an atom if data is provided

Calculating Mass Number

The mass number can be calculated if the number of protons and neutrons are known:

Mass Number (A) = Number of Protons + Number of Neutrons

Alternatively, for a specific isotope, the notation is written as:

Element symbol with mass number: ^{12}C (Carbon-12)

where 12 indicates the mass number.

Isotopes and Their Significance in Chemistry

What are Isotopes?

Isotopes are variants of a particular chemical element that have the same number of protons but different numbers of neutrons. This variation results in different mass numbers but similar chemical properties.

- Examples: Carbon-12 (^{12}C), Carbon-13 (^{13}C), and Carbon-14 (^{14}C).
- Some isotopes are stable, while others are radioactive.

Importance of Isotopes

Isotopes have practical applications in various fields:

- Radioactive dating (e.g., Carbon-14 dating in archaeology)
- Medical imaging and treatment (e.g., Radioisotopes in cancer therapy)
- Tracing chemical reactions in research
- Understanding atomic structure and nuclear physics

Practice Exercises for Students

To reinforce learning, a chemistry worksheet should include practice questions such as:

- Identify the atomic number and mass number of an element given its isotope notation (e.g., $^{40}_{20}\text{Ca}$).
- Calculate the number of neutrons in the isotope ^7_3Li , given that Lithium's atomic number is 3.
- Write the isotope notation for an atom with 11 protons and 12 neutrons.
- Determine whether a given isotope is stable or radioactive based on its atomic and mass numbers.
- Compare two isotopes of the same element and discuss how their properties differ.

Answers to these questions help students develop confidence and deepen their understanding of atomic concepts.

Visual Aids and Diagrams in the Worksheet

Incorporating visual elements enhances comprehension:

- Diagrams of atomic structure highlighting protons, neutrons, and electrons.
- Periodic table with atomic numbers and isotopic information.
- Charts showing common isotopes of key elements.
- Visual comparisons between isotopes to illustrate differences in neutrons.

Such visuals make abstract concepts more tangible and facilitate better retention.

Additional Resources and Study Tips

Students seeking to excel in understanding atomic number and mass number should consider:

- Using periodic tables effectively, noting atomic numbers and isotopic masses.
- Practicing calculations with real data from elements and isotopes.
- Engaging in hands-on activities, such as building atom models.

- Utilizing online interactive quizzes and simulations for practice.

Consistent practice with a variety of problems enhances mastery and prepares students for higher-level chemistry topics.

Conclusion

A comprehensive chemistry worksheet on atomic number and mass number is an invaluable educational resource that reinforces core atomic concepts vital for understanding the nature of matter. By exploring definitions, differences, calculations, and practical applications, learners gain a solid foundation in atomic structure. Integrating visual aids, practice exercises, and real-world examples within the worksheet makes learning engaging and effective. Mastery of these concepts not only prepares students for exams but also deepens their appreciation of the intricate building blocks of the universe, paving the way for success in advanced chemistry and related sciences.

Chemistry Worksheet Atomic Number and Mass Number: An In-Depth Exploration

In the realm of chemistry, understanding the fundamental building blocks of matter is essential for grasping how elements interact, form compounds, and underpin the universe itself. Among the foundational concepts are the atomic number and mass number, which serve as key identifiers and descriptors of atoms. These numerical values not only facilitate the classification of elements but also underpin various calculations and models within atomic physics and chemistry. This article offers a comprehensive review of these concepts, exploring their definitions, significance, differences, and applications, while providing an analytical perspective on their roles in chemical education and research.

Understanding Atomic Number and Mass Number: Definitions and Significance

What is the Atomic Number?

The atomic number, symbolized as Z , is a fundamental property of an element that specifies the number of protons in the nucleus of an atom. Since protons carry a positive charge, the atomic number directly determines the element's identity. For example, all atoms with an atomic number of 6 are carbon atoms, while those with an atomic number of 8 are oxygen atoms.

Significance of the Atomic Number:

- Element Identification: The atomic number uniquely identifies an element; no two different elements share the same atomic number.

- Periodic Table Organization: Elements are arranged in order of increasing atomic number, which structures the periodic table and reflects periodic trends.
- Chemical Properties: The atomic number influences an atom's electron configuration, dictating its chemical reactivity, bonding behavior, and valency.

What is the Mass Number?

The mass number, denoted as A, represents the total number of nucleons (protons and neutrons) in an atom's nucleus. Unlike the atomic number, which is fixed for each element, the mass number can vary within the same element, leading to different isotopes.

Significance of the Mass Number:

- Isotope Identification: Variations in the mass number give rise to isotopes, which are atoms of the same element with different neutron counts.
- Mass Calculations: The mass number is used to approximate the atomic mass of an isotope, which is vital for calculations in stoichiometry and nuclear chemistry.
- Radioactivity and Stability: The ratio of neutrons to protons (reflected in the mass number) influences nuclear stability and the likelihood of radioactive decay.

Distinguishing Between Atomic Number and Mass Number: Key Differences

Understanding the differences between atomic number and mass number is crucial for students and professionals alike. Below are the primary distinctions:

Aspect	Atomic Number (Z)	Mass Number (A)
Definition	Number of protons in the nucleus	Total number of protons and neutrons in the nucleus
Units	No units; a count	No units; a count
Fixed for an element	Yes	No; varies with isotopes
Determines	Element identity	Isotope identity
Affects	Chemical properties	Nuclear properties and atomic mass

Additional insights:

- Since the atomic number determines the element, changing Z results in a different element altogether.
- The mass number can vary within the same element, leading to different isotopes with distinct physical properties.

Calculating Atomic Mass and Isotopic Composition

The Role of Isotopes in Atomic Mass Calculations

While the atomic mass listed on the periodic table is a weighted average, individual isotopes have specific masses approximately equal to their mass numbers. Understanding isotopic composition is essential for applications in:

- Radiometric dating
- Medical imaging
- Nuclear energy

Calculating Atomic Mass:

The atomic mass of an element is calculated as the weighted average of the masses of its naturally occurring isotopes:

\[

$$\text{Atomic mass} = \sum (\text{fractional abundance of isotope} \times \text{mass of isotope})$$

\]

For example, chlorine has two main isotopes:

- Chlorine-35 (mass number 35, abundance ~75%)
- Chlorine-37 (mass number 37, abundance ~25%)

The average atomic mass:

\[

$$(0.75 \times 35) + (0.25 \times 37) = 26.25 + 9.25 = 35.5 \text{ amu}$$

\\

This explains why chlorine's atomic mass appears as approximately 35.5 amu on the periodic table.

Implications for Chemical and Physical Properties

Isotopic differences influence physical properties such as:

- Atomic weight
- Spectral lines
- Nuclear stability

In chemical reactions, isotopic substitution can lead to subtle effects, such as isotope effects, affecting reaction rates and mechanisms.

Using Atomic Number and Mass Number in Practice

Notation and Representation

Atoms are commonly represented using nuclear notation:

\\

$$^A_Z\text{X}$$

\\

Where:

- X is the chemical symbol
- Z is the atomic number
- A is the mass number

For example, an isotope of carbon with 6 protons and 7 neutrons is written as:

\\

$^{13}_6\text{C}$ $\backslash$

This notation succinctly conveys the isotope's identity and composition.

Determining Neutron Number

Given the atomic number and mass number, the neutron count can be calculated:

 $\backslash$ $\text{Number of neutrons} = A - Z$ $\backslash$

Using the previous example:

 $\backslash$ $7 = 13 - 6$ $\backslash$

Understanding neutron numbers is critical in analyzing nuclear stability and decay processes.

Applications in Radioactive Decay and Nuclear Reactions

- Radioactive isotopes are characterized by unstable nuclei with specific neutron-to-proton ratios.
- Nuclear reactions often involve changes in the mass and atomic numbers, such as in nuclear fission or fusion.
- Detection and measurement of isotopes rely on their unique atomic and mass numbers.

Educational and Practical Importance of the Worksheet Approach

Using worksheets to teach atomic number and mass number enhances comprehension through active engagement. They often include exercises such as:

- Filling in atomic and mass numbers for different isotopes

- Calculating neutrons and protons
- Converting between notation forms
- Interpreting atomic data for element identification

Benefits of such worksheets include:

- Reinforcing theoretical knowledge
- Developing problem-solving skills
- Preparing students for laboratory and exam scenarios
- Facilitating deeper understanding of atomic structure

Advanced Topics and Contemporary Research

Beyond foundational concepts, modern research in nuclear physics and chemistry explores:

- Superheavy elements: Atoms with extremely high atomic and mass numbers, often synthesized in laboratories.
- Isotope enrichment: Techniques to increase the proportion of specific isotopes for medical or industrial use.
- Nuclear medicine: Utilization of radioactive isotopes (e.g., technetium-99m) for diagnostics and treatment.
- Nuclear stability theories: Models predicting the stability and decay pathways based on neutron-proton ratios.

These advanced topics demonstrate how the understanding of atomic and mass numbers continues to evolve, impacting technology and medicine.

Conclusion: The Integral Role of Atomic and Mass Numbers in Chemistry

The concepts of atomic number and mass number are cornerstones of atomic theory and chemistry. They serve as precise identifiers for elements and isotopes, underpin calculations in stoichiometry, and inform our understanding of nuclear phenomena. As educational tools, worksheets that focus on these numbers cultivate critical thinking and foundational knowledge, essential for any aspiring chemist, physicist, or researcher. Recognizing their distinctions and applications enhances our comprehension of matter at its most fundamental level and underscores the intricate beauty of the atomic world.

In summary, mastery of atomic and mass numbers is not merely academic but vital for advancing

scientific inquiry, technological innovation, and practical applications across multiple disciplines.

QuestionAnswer What is the atomic number of an element? The atomic number of an element is the number of protons found in the nucleus of an atom of that element. How is the mass number of an atom different from its atomic number? The mass number is the total number of protons and neutrons in an atom's nucleus, whereas the atomic number only counts the protons. Why is the atomic number important in identifying elements? Because each element has a unique atomic number, it serves as the primary identifier for distinguishing one element from another. Can the mass number of an atom change? Yes, the mass number can change due to the addition or removal of neutrons, resulting in different isotopes of the same element. How do you calculate the atomic number and mass number from an element's symbol? The atomic number is usually written as a subscript before the symbol, and the mass number as a superscript. For example, in ${}^2_1\text{H}$, 1 is the atomic number and 2 is the mass number. What is an isotope, and how is it related to atomic and mass numbers? An isotope is an atom of the same element with the same atomic number but a different mass number due to varying neutrons. Why are isotopes of an element chemically similar? Because isotopes have the same number of protons and electrons, they exhibit similar chemical behavior despite differences in mass. How can understanding atomic and mass numbers help in scientific studies? It helps in identifying elements and their isotopes, understanding nuclear reactions, and calculating atomic weights for chemical calculations.

Related keywords: atomic number, mass number, isotopes, protons, neutrons, electrons, atomic structure, periodic table, atomic mass, element symbols

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